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~~Quantum Numbers~~

Note Taking Guide: Episode 304 Name Aditi Acharya Quantum Numbers n, l, m, s Used to describe an electron in an atom

n Principle quantum number Represents main energy level of electron maximum # of electrons in an Energy level = $2n^2$

Example: What is the maximum number of electrons that can be in the 5th main energy level? 1 The 2nd quantum number l Describes the orbital shape within an energy level l number of orbital shapes possible in energy level = n Orbital Shapes designated s, p ...

~~note_ep304aditi.pdf orientation of orbital in space s ...~~

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of _electron _ in an energy level _ = $2n^2$ _ Example: What is the maximum number of electrons that can be in the 5th _ main energy level? 1 The 2 nd quantum number _____ Describes the orbital _____ shape within an ...

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gravity = Example problems: Rhoda Tiller
looks up to see a flowerpot fall off a ledge
9.2 m above her head.

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